

# Synthesis of [ $^{14}\text{C}$ ] labeled 2-methoxypyrimidine-5-carboxylic acid

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**A versatile method for  $^{14}\text{C}$  labeling of 2-methoxypyrimidine-5-carboxylic acid at the 2-position has been developed after encountering difficulties with traditional approaches to label the carboxyl function. The method developed can also be used for  $^{14}\text{C}$  labeling other positions of the pyrimidine ring system.**

**Keywords:** [ $^{14}\text{C}$ ] 2-methoxypyrimidine-5-carboxylic acid; [ $^{14}\text{C}$ ] O-methylisourea hydrochloride

## Introduction

The syntheses of 2-methoxypyrimidine-5-carboxylic acids with a  $^{14}\text{C}$  label at the 2-position of the pyrimidine ring **2**, or at 4-, 5- and 6-positions of **3** involve multi-step process compared to the simple introduction of  $^{14}\text{C}$  on the carboxyl function of **1** (Figure 1). This report discusses our initial attempts to synthesize carboxyl labeled 2-methoxypyrimidine-5-carboxylic acid and our eventual success in preparing ring labeled title compound.

## Results and discussion

Labelling of carboxyl function of **1** was attempted by employing known methods such as metal-halogen exchange/Grignard reactions<sup>1–2</sup> and treatment with [ $^{14}\text{C}$ ]  $\text{CO}_2$ . None of these methods resulted in the synthesis of desired  $^{14}\text{C}$  labeled carboxylic acid derivative. Thus, when 5-bromo-2-methoxypyrimidine was reacted with *n*-BuLi followed by [ $^{14}\text{C}$ ]  $\text{CO}_2$  at  $-78$  to  $-20^\circ\text{C}$ , the reaction gave an unidentified product along with unreacted starting material.

Aromatic nitriles are well-known precursors for the corresponding carboxylic acids.<sup>3</sup> Aryl nitriles have been prepared by cyanation of iodides or bromides using CuCN or KCN in absence of solvent or in solvents like pyridine, quinoline, DMF, NMP at elevated temperature.<sup>4–11</sup> Under Von Braun conditions the reaction of 5-bromo-2-methoxypyrimidine with [ $^{14}\text{C}$ ] CuCN in DMF at  $145^\circ\text{C}$  for 48 h did not yield cyano derivative.

Palladium catalyzed cyanations of aryl bromides promoted by organotin compounds have been successfully used for the preparation of a number of aryl cyanides including 3-cyanopyrimidine and 3-cyanoquinolines.<sup>12</sup> However, reaction of 5-bromo-2-methoxypyrimidine with [ $^{14}\text{C}$ ] KCN under these conditions resulted only in the recovery of starting material.

The synthesis of 2-methoxypyrimidine-5-carboxylic acid **2** with a label at 2-position of pyrimidine ring is an alternative choice. A number of synthetic methods are reported in the literature for pyrimidine 5-carboxylic acid derivatives.<sup>13–16</sup> However, a few direct approaches leading to pyrimidine ring that lacks substitution at 4-position have been reported.<sup>17–19</sup>

Our approach to compound **2** is as depicted in Scheme 1. [ $^{14}\text{C}$ ] barium cyanamide (**4**) was converted to [ $^{14}\text{C}$ ] cyanamide (**5**) by the treatment with aqueous  $\text{H}_2\text{SO}_4$  in 75% yield. Reaction of [ $^{14}\text{C}$ ] cyanamide (**5**) with anhydrous  $\text{CH}_3\text{OH}$  in the presence of dry HCl gas for 3 days according to the literature method<sup>20</sup> gave [ $^{14}\text{C}$ ] O-methylisourea hydrochloride (**6**).

In parallel, the sodium salt of 3,3-dimethoxy-2-methoxycarbonylpropen-1-ol (**7**) was synthesized by the condensation of methyl formate with methyl 3,3-dimethoxypropionate in the presence of NaH as described in the literature.<sup>19</sup> The sodium salt **7** was reacted with [ $^{14}\text{C}$ ] O-methylisourea hydrochloride (**6**) in DMF at  $100^\circ\text{C}$  to give [ $2\text{-}^{14}\text{C}$ ] methyl 2-methoxypyrimidine-5-carboxylate (**8**) in 70% yield, identified by  $^1\text{H}$  NMR. The hydrolysis of ester **8** was carried out by heating with 2 N NaOH in aqueous dioxane.<sup>21</sup> [ $2\text{-}^{14}\text{C}$ ] 2-methoxypyrimidine-5-carboxylic acid (**2**) was isolated after acidification of the mixture with 2 N HCl in 97% yield and radiochemical purity of 98.41% by HPLC. Specific activity was determined to be 56.4 mCi/mmol. The overall yield of [ $2\text{-}^{14}\text{C}$ ] 2-methoxypyrimidine-5-carboxylic acid was 13.7% from **4**.

This methodology led to a successful labeling in the 2-position. This approach may also be applied to other pyrimidine carboxylic acid derivatives **3** with  $^{14}\text{C}$  labeling at the 4-, 5- and/or 6-position by using appropriately labeled methyl formate or methyl 3,3-dimethoxypropionate. Labeling can also be extended to the carboxyl moiety as well using this scheme starting from **7** labeled at the ester carbon.

## Experimental

All reagents and solvents were purchased from Sigma-Aldrich Chemical Company. [ $^{14}\text{C}$ ] Barium cyanamide was obtained from

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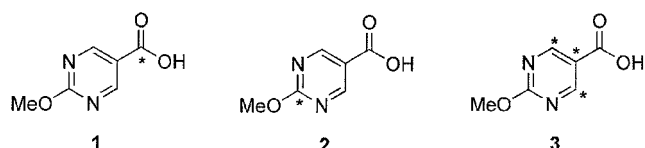
I.U.T. GmbH, Germany.  $^1\text{H}$  NMR spectra were recorded on a Bruker 300 MHz spectrometer. Internal TMS was used as a reference standard. The final product was identified by HPLC comparison with commercially available material on a Zorbax SB C-18 column using (A) 0.1% TFA in  $\text{H}_2\text{O}$  (B)  $\text{CH}_3\text{CN}$ , 0–50% B linear gradient in 25 min, holding 50% B for 5 min, 1 ml/min, UV 254 nm.

#### [ $^{14}\text{C}$ ] cyanamide (5)

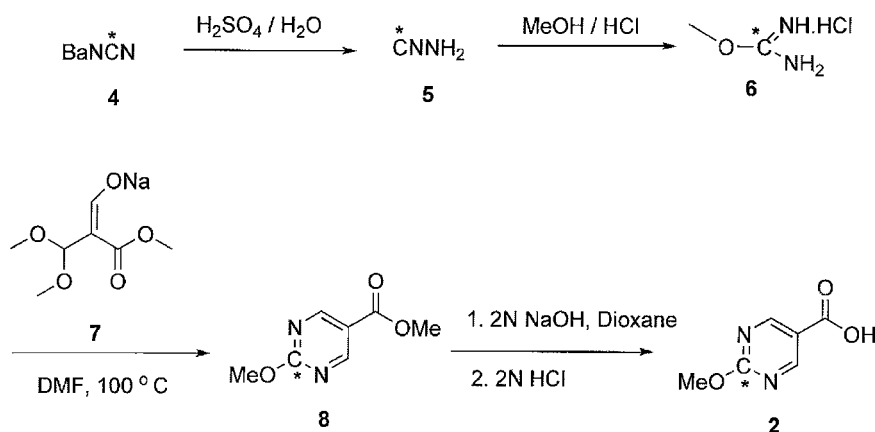
To a suspension of [ $^{14}\text{C}$ ] barium cyanamide (**4**) (1200 mCi, specific activity 57.6 mCi/mmol, 20.8 mmol) in  $\text{H}_2\text{O}$  (14 ml) cooled to 0–5°C, was added conc.  $\text{H}_2\text{SO}_4$  (1.05 ml, 20.8 mmol) drop wise during about 15 min. The resulting white suspension was stirred at 0–5°C for 1 h and centrifuged. The supernatant liquid was decanted and the residue stirred with  $\text{H}_2\text{O}$  (10 ml) and centrifuged. The process was repeated and the combined supernatant was extracted with EtOAc (10  $\times$  30 ml). The organic extract was dried over  $\text{Na}_2\text{SO}_4$ , filtered and the solvent was removed in vacuum to yield a colorless solid (904 mCi, 75.3%). The product co-chromatographed with standard by radio-TLC on avicel plate using *n*-BuOH:  $\text{NH}_4\text{OH}$ :  $\text{H}_2\text{O}$  (12:3:5) as solvent system.

#### [ $^{14}\text{C}$ ] O-methylisourea hydrochloride (6)

Anhydrous HCl gas was bubbled through a solution of **5** (335 mCi, 5.82 mmol) in anhydrous  $\text{CH}_3\text{OH}$  (5.0 ml) for 15 min. The reaction flask containing colorless solid was sealed and set aside at ambient temperature for 3 days. The solvent was removed under reduced pressure and the solid obtained was dried in a vacuum desiccator over  $\text{P}_2\text{O}_5$  and KOH overnight. The [ $^{14}\text{C}$ ] O-methylisourea hydrochloride (**6**) obtained (0.46 g, 242 mCi, 72.2%) was used in the next reaction without further purification.  $^1\text{H}$  NMR ( $\text{DMSO}-d_6$ ):  $\delta$  8.7 (bs,  $\text{NH}_2$ ), 3.96 (s, 3H,  $\text{CH}_3$ ).



**Figure 1.** 2-Methoxypyrimidine-5-carboxylic acids with [ $^{14}\text{C}$ ] label at different positions.



**Scheme 1.** Synthesis of [ $^{14}\text{C}$ ] 2-methoxypyrimidine-5-carboxylic acid.

#### [2- $^{14}\text{C}$ ] methyl 2-methoxypyrimidine-5-carboxylate (8)

A mixture of **6** (242 mCi, 4.20 mmol) and sodium 3,3-dimethoxy-2-carbomethoxyprop-1-en-1-oxide (**7**)<sup>19</sup> (0.96 g, 4.83 mmol) in DMF (8.0 ml) was heated to 110°C and maintained for 2 h. The reaction mixture was cooled to ambient temperature and  $\text{H}_2\text{O}$  (30.0 ml) was added. The solid separated was extracted with  $\text{CH}_2\text{Cl}_2$  (3  $\times$  30.0 ml). The combined organic extracts were dried over  $\text{MgSO}_4$  and filtered. The filtrate (214 mCi) was concentrated under vacuum to yield a yellow solid. Radio-TLC analysis on silica gel plate [hexane: EtOAc (1:1)] indicated about 93% of the product co-eluting with the standard. The crude material was purified by a silica gel flash chromatography using hexane: EtOAc (9:1) as the eluent. Homogenous fractions were pooled and the solvent was removed under vacuum to yield [ $^{14}\text{C}$ ] methyl 2-methoxypyrimidine-5-carboxylate (**8**) as a colorless solid (170 mCi, 70.3%).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  9.08 (s, 2H, H-4 and 6), 4.09 (s, 3H,  $\text{CH}_3$ ) and 3.9 (s, 3H,  $\text{CH}_3$ )

#### [2- $^{14}\text{C}$ ] 2-methoxypyrimidine-5-carboxylic acid (2)

To a solution of **8** (170 mCi, 2.95 mmol) in a mixture of dioxane:  $\text{H}_2\text{O}$  (1:1, 34.0 ml) was added 2N NaOH (1.77 ml, 3.54 mmol) and stirred at ambient temperature overnight. The clear solution was concentrated under reduced pressure to  $\sim$ 1 ml and diluted with  $\text{H}_2\text{O}$  (10.0 ml). Resulting turbid aqueous solution was washed with  $\text{CHCl}_3$  (10.0 ml) and acidified with 2N HCl to pH 2. Colorless solid was filtered, washed sequentially with  $\text{H}_2\text{O}$  (2  $\times$  2.0 ml),  $\text{H}_2\text{O}$ :  $\text{CH}_3\text{CN}$  (1:1, 2  $\times$  2.0 ml) and dried to a constant weight. The product obtained was identified as [ $^{14}\text{C}$ ] 2-methoxypyrimidine-5-carboxylic acid (**2**) (0.45 g, 96.84%) with a radiochemical purity of 98.4% by HPLC.  $^1\text{H}$  NMR ( $\text{DMSO}-d_6$ ):  $\delta$  9.03 (s, 2H, H- at 4 and 6), 4.00 (s, 3H,  $\text{CH}_3$ ), specific activity: 56.4 mCi/mmol determined by weight assay.

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## References

- [1] A. Murray III, W. H. Langham, *J. Am. Chem. Soc.* **1952**, *74*, 6289–6295.
- [2] V. Grignard, *Ann. Chim.* **1901**, *24*, 433–490.
- [3] C. A. Buehler, D. E. Pearson, *Survey of Organic Synthesis*, chapter 13, Wiley Interscience, New York, 1970, pp. 752–755 and references cited therein.
- [4] A. Pongratz, *Monatsh. Chem.* **1927**, *48*, 585.
- [5] A. Pongratz, *Monatsh. Chem.* **1929**, *52*, 7.
- [6] K. W. Rosenmund, E. Struck, *Chem. Ber.* **1919**, *52*, 1749–1756.
- [7] J. von Braun, G. Manz, *Liebigs Ann. Chem.* **1931**, *488*, 111–126.
- [8] D. T. Mowry, *Chem. Rev.* **1948**, *42*, 189–283.
- [9] K. Fiedrich, K. Wallenfels, in: *The Chemistry of the Cyano Group* (Ed.: Z. Rappoport), Interscience, London, 1970, p. 67.
- [10] L. Friedman, H. Shechter, *J. Org. Chem.* **1961**, *26*, 2522–2524.
- [11] M. S. Newman, H. Boden, *J. Org. Chem.* **1961**, *26*, 2525.
- [12] C. Yang, J. M. Williams, *Org. Lett.* **2004**, *6*, 2837–2840.
- [13] R. Urban, O. Schnider, *Helv. Chim. Acta* **1958**, *41*, 1806.
- [14] S. Kohra, Y. Tominaga, A. Hosomi, *J. Heterocycl. Chem.* **1988**, *25*, 959–968.
- [15] A. Lorente, L. Vaquerizo, A. Martin, P. Gomez-Sal, *Heterocycles* **1995**, *41*, 71–86.
- [16] A. Guzman, M. Romero, F. X. Talamas, R. Villena, R. Greenhouse, J. M. Muchowski, *J. Org. Chem.* **1996**, *61*, 2470–2483.
- [17] P. Schenone, L. Sansebastiano, L. Mosti, *J. Heterocycl. Chem.* **1990**, *27*, 295–305.
- [18] E. Dyer, T. B. Johnson, *J. Am. Chem. Soc.* **1934**, *56*, 222–225.
- [19] P. Zhichkin, D. J. Fairfax, S. A. Eisenbeis, *Synthesis* **2002**, 720–722.
- [20] F. Kurzer, A. Lawson, *Organic Syntheses*, Vol. 4, Wiley, New York, 1963, pp. 645–649.
- [21] G. Brooks, E. Hunt, S. Howard, US Patent 0114674, 2003.